

Ground Testing of the JPL Electronic Nose: Mechanisms of Response in Conductometric Polymer-Carbon Composite Sensors

M. A. Ryan, M. L. Homer, A. Manfreda, A. D. Jewell, L. Lara,
S.P.S Yen, H. Zhou, B. Lin, A. Shevade, A.K. Kisor and J. Lim

Jet Propulsion Laboratory
California Institute of Technology
Pasadena CA 91109





AIR QUALITY MONITORING REQUIREMENTS

Monitor cabin environment with time stamped measurements and rates of change

- ◆ Major constituents - "near-continuous" monitoring
 - N₂ O₂ CO₂ H₂O
 - H₂ CH₄ CO
- ◆ Trace contaminants - less frequent monitoring required - 40 organic compounds
- ◆ Event Monitoring - rapid response time
 - ◆ pyrolysis markers - CO HCN HCl
 - ◆ marker compounds for electronics overheating
 - ◆ monitoring for sudden release from fluid systems, experiments, EVA, waste
 - ◆ follow progress of decontamination after an event

based on AEMC Requirements for Technology Development, 1998, Section 5

AIR QUALITY MONITORING

JPL ELECTRONIC NOSE

FUNCTIONS

- ◆ Incident monitor for targeted contaminants exceeding Spacecraft Max Allowable Concentration (SMAC). Identify and quantify target compounds at SMAC level.
- ◆ Monitor for presence of compounds associated with fires or overheating electronics
- ◆ Monitor clean-up process

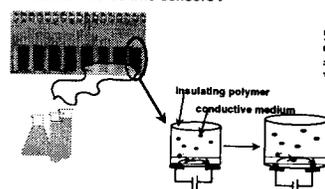
CHARACTERISTICS

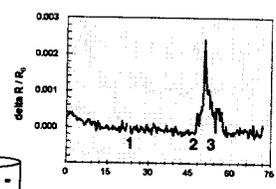
- ◆ Low mass, low power device
 - First Generation 1.5 kg; ~ 2 L; 1.5 W avg, 3 W peak
 - Second Generation < 1 kg; < 1 L; 1.5 W avg, 3 W peak
- ◆ Requires little crew time for maintenance and calibration

WHAT IS AN ELECTRONIC NOSE?

An array of non-specific chemical sensors, controlled and analyzed electronically, which have overlapping responses to compounds. Compounds are identified and quantified by recognition of patterns of response.

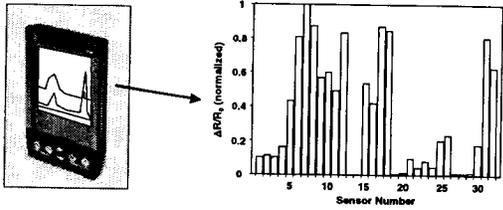
1. ENose measures background resistance in each sensor and establishes R₀ (baseline).
2. Contaminant comes in contact with and sorbs into sensors.
3. The sensors, polymer films loaded with a conductive medium such as carbon black, change resistance by swelling or shrinking as air composition changes.





WHAT IS AN ELECTRONIC NOSE?

4. Resistance is recorded, the change in resistance is computed, and the distributed response pattern ("fingerprint") of the sensor array is used to identify gases and mixtures of gases. The magnitude of response is used to quantify the identified compound.



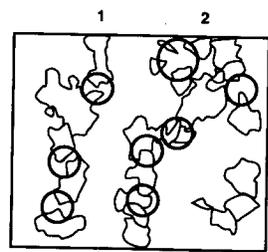
The diagram shows a handheld electronic nose device on the left with a screen displaying a waveform. An arrow points from the device to a bar chart on the right. The bar chart plots $\Delta R/R_0$ (normalized) on the y-axis (ranging from 0 to 1) against Sensor Number on the x-axis (ranging from 0 to 30). The bars show a characteristic pattern of peaks and troughs.

5. Responses of the sensor array are analyzed and quantified using software developed for the task.

n-propanol, 30 ppm

CONDUCTION IN POLYMER-CARBON COMPOSITES

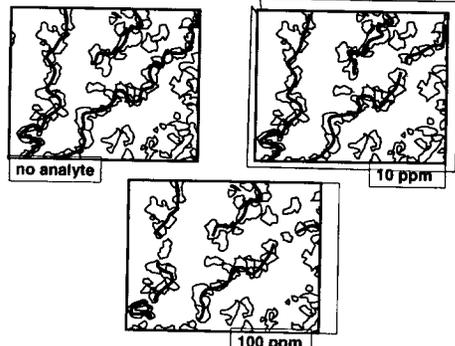
- Tunneling**
 - Carbon to carbon through polymer matrix
 - Resistance is controlled by electric field between adjacent particles
- Percolation Controlled**
 - Carbon-to-carbon through particle contact
 - Resistance is controlled by contact resistance between particles and by gaps too large for tunneling



The diagram shows two scenarios of carbon particles (represented as circles) in a polymer matrix. Scenario 1 shows particles in close proximity, with arrows indicating electron tunneling between them. Scenario 2 shows particles that are further apart, with gaps between them that are too large for tunneling to occur, illustrating percolation control.

CONDUCTION IN POLYMER-CARBON COMPOSITES

Concentration of analyte determines degree of swelling, which determines magnitude of response

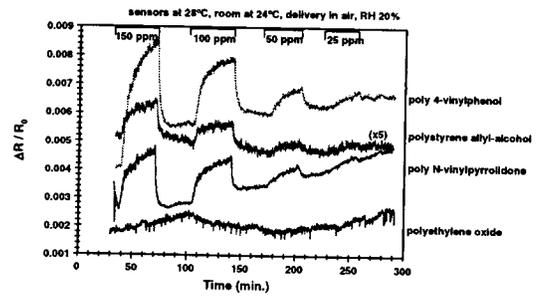


The diagram illustrates three stages of polymer swelling in a carbon composite. The top-left panel, labeled "no analyte", shows a relatively compact polymer network. The top-right panel, labeled "10 ppm", shows the polymer network beginning to swell and expand. The bottom panel, labeled "100 ppm", shows the polymer network significantly swollen and expanded, which increases the contact between carbon particles and thus the conductivity.

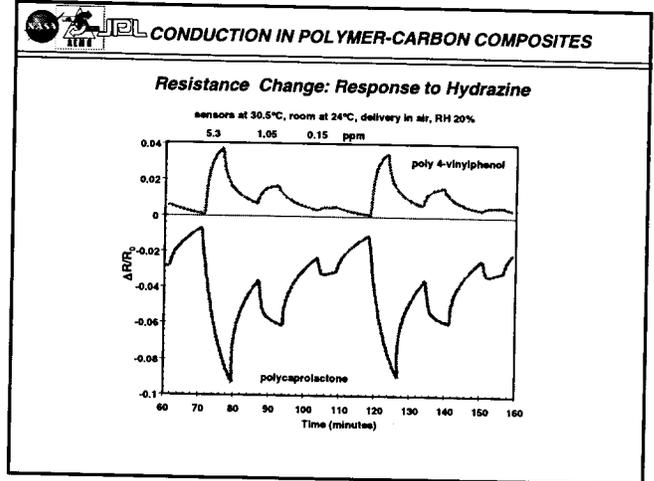
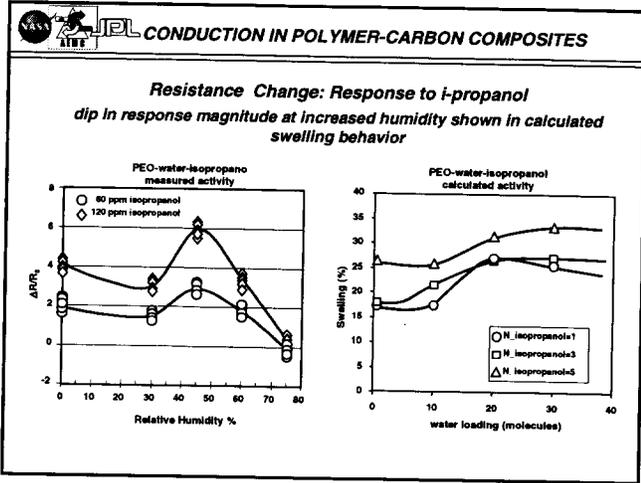
CONDUCTION IN POLYMER-CARBON COMPOSITES

Resistance Change: Response to Methanol

sensors at 23°C, room at 24°C, delivery in air, RH 20%



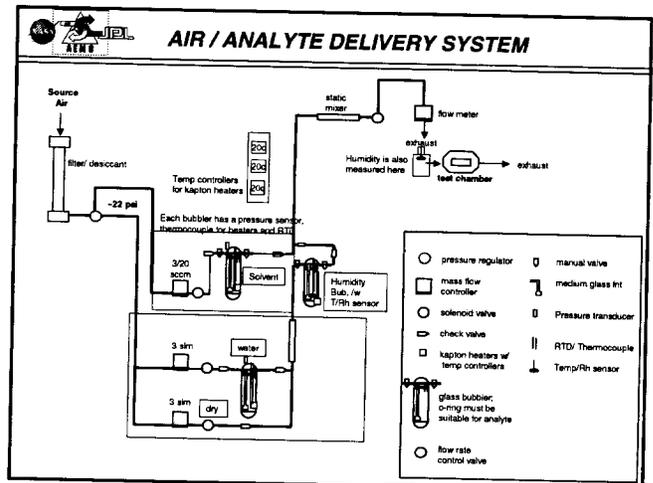
The graph plots $\Delta R/R_0$ on the y-axis (ranging from 0.001 to 0.009) against Time (min.) on the x-axis (ranging from 0 to 300). Four curves are shown, corresponding to different polymer matrices: poly 4-vinylphenol, polystyrene allyl-alcohol (x5), poly N-vinylpyrrolidone, and polyethylene oxide. Each curve shows a step-like increase in resistance when methanol is introduced at 50, 100, 150, and 200 ppm, followed by a gradual decrease back to baseline. The magnitude of the resistance change increases with the concentration of methanol.



AIR / ANALYTE DELIVERY SYSTEM

A computer-controlled gas delivery system bubbles air through solvents and mixes them with humidified air. Temperature is controlled in the gas delivery system using heaters.

- concentrations are verified using a Rosemont Hydrocarbon Analyzer calibrated to standard gases (methanol, methane, toluene, acetone). Humidity is verified using calibrated sensors.
- Background gases can be cleaned air or bottled gas. Humidity is added by bubbling through water and mixing humidified air with dry air.

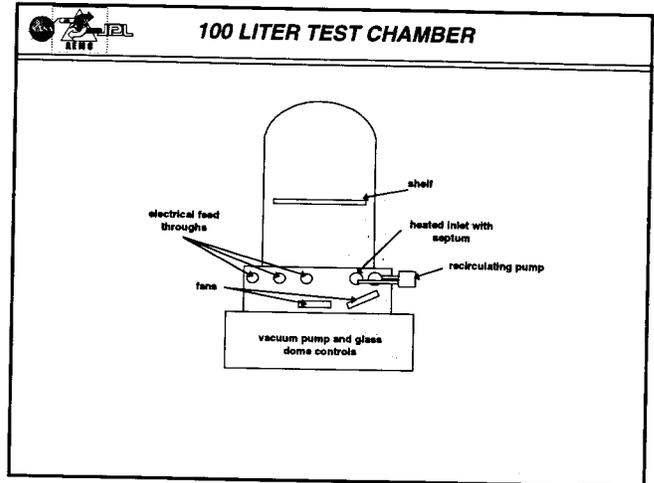


100 LITER TEST CHAMBER

DEVICE OPERATION AND TESTING

Test Chamber

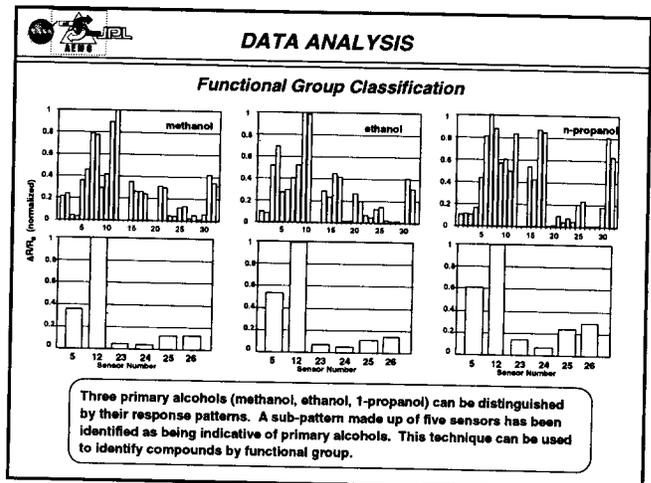
- ◆ 100 L test chamber
- ◆ Temperature controlled using a water jacket
- ◆ Pressure controlled by evacuating and backfilling with clean, humidified air
- ◆ Test analytes are delivered using a syringe for rapid injection (to simulate a spill) or a syringe pump (to simulate a leak).
- ◆ Solid analytes (indole) are delivered by placing a measured quantity in a feedthrough. Gaseous analytes (burning insulation) is delivered through a tube.
- ◆ Chamber air can be circulated within the chamber or directed through a filter to simulate the Air Revitalization System.
- ◆ Testing will begin after a complete training set has been taken. Data analysis is automated to find events and identify them.



100 LITER TEST CHAMBER

Temperature and Pressure Controlled 100 L Test Chamber

- ◆ Liquid test analytes are delivered using a syringe or a syringe pump
- ◆ Solid analytes are delivered by placing a measured quantity in a feedthrough; gaseous analytes are delivered through a tube.
- ◆ Chamber air can be circulated within the chamber or directed through a filter to simulate the Air Revitalization System.



DATA ANALYSIS

Functional Group Classification

Classify Unknowns In Mixtures and/or With More Than One Functional Group

preliminary software was for single unknowns, with a single functional group. focus was on alcohols and aromatics.

developed software routine for choosing best subsets of sensors for functional group classification based on fingerprint group distance calculation. Software will find a subset of sensors that has a minimum value of the distance between one particular group divided by the distance between the rest of the target analytes. Each different functional group has a different optimal subset of sensors, and so do different combinations of functional groups. This sensor selection software is a very useful tool and may be applicable to other general sensor selection /evaluation needs.

Also devised a two-step data analysis approach in which the result of the first (main) step will be given preference in the second (functional group analysis) step. For example, if the first step results in "methanol" plus "freon" with an unacceptably large residual (r_0), then the software will assume it could be of some unknown analytes.

DATA ANALYSIS

Functional Group Classification

trained analytes	untrained	mixture events were generated (simulated) from one untrained analyte plus one trained or untrained analytes Total tested are 171 different combinations of binary mixtures; success is correct functional group classification
methanol	2-propanol	
ethanol	butanol	
1-propanol	pentanol	
methane	phenol	
ammonia	anthracene	
benzene	ethyl benzene	
formaldehyde	p-xylene	
freon 113	o-xylene	
indole	m-xylene	
toluene),		

success rates	
2-propanol	86%
butanol	67%
pentanol	76%
phenol	67%
anthracene	86%
ethyl benzene	95%
p-xylene	95%
o-xylene	86%
m-xylene	100%

Analyses for 2nd Generation Enose

	tested limits				
	24 hr SMAC	1/3 SMAC	3x SMAC	low high	
acetaldehyde (1500ppm gas)	6	2.00	18	2	20
acetone	200	66.67	600	65	600
acetonitrile (1500 ppm gas)	4	1.33	12	1	25
ammonia - NH ₃	20	6.67	60	6	60
benzene (1510 ppm gas)	3	1.00	9	3	75
2-butanone	50	16.67	150	15	150
chlorobenzene	10	3.33	30	3	30
dichloromethane	35	11.67	105	11	150
dichloroethane	0.4	0.13	1.2	TBD	TBD
ethanol	2000	666.67	6000	665	6000
ethyl benzene	60	20.00	180	18	165
formaldehyde	0.1	0.03	0.3	TBD	TBD
freon 113	50	16.67	150	15	150
furan (500 ppm gas)	0.36	0.12	1.08	TBD	TBD
hexane	50	16.67	150	15	150
hexamethyltricycloallane	25	8.33	75	TBD	TBD
hydrazine	0.3	0.10	0.9	TBD	TBD
indole	0.3	0.10	0.9	TBD	TBD
methane	6300	1766.67	15900	3000	40000
methanol	10	3.33	30	3	100
methyl hydrazine	0.002	0.0007	0.006	TBD	TBD
iso-propanol	100	33.33	300	30	600
tetrahydrofuran	40	13.33	120	13	120
toluene	16	5.33	48	5	50
1,1,1-trichloroethane	11	3.67	33	5	200
o,p-xylene (use m value)	100	33.33	300	33	300

ACKNOWLEDGEMENTS

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