



# Advances in Materials for PEM Fuel Cells and Electrolyzers for Exploration System Missions

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# H<sub>2</sub>/O<sub>2</sub> Primary and Regenerative Fuel Cells

- Primary Fuel Cell (H<sub>2</sub>/O<sub>2</sub> → electricity)
- Electrolyzer (electricity → H<sub>2</sub>/O<sub>2</sub>)
- Applications
  - Mobile Power sources
    - Automobiles, Lunar Rovers, Lunar Landers, Unmanned Aircraft
  - Large scale energy storage
    - Renewable Energy Storage, Lunar Surface Systems
  - “Back-up” Power and Battery Recharging
    - Remote Cell Phone Towers, Materials Handling Equipment



# Challenges

<b>Application</b>	<b>Barriers</b>
Automobiles	Materials <u>Cost</u> and <u>Durability</u>
Lunar Systems	<u>Efficiency</u> and Durability and technology maturity
Unmanned Aircraft	<u>Fuel Storage</u> , specific Energy
Renewable Energy Storage	Cost, efficiency, and storage logistics
Back up Power	Cost and durability

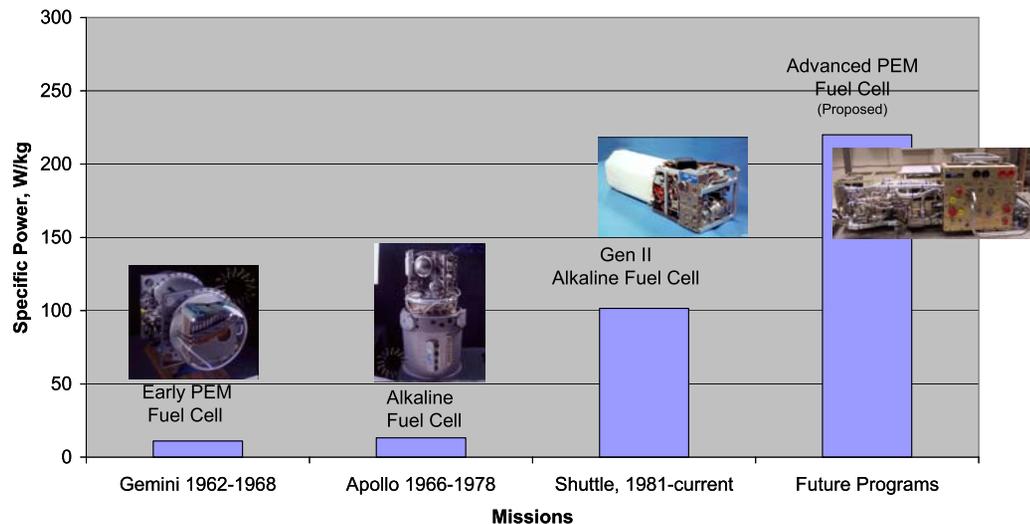
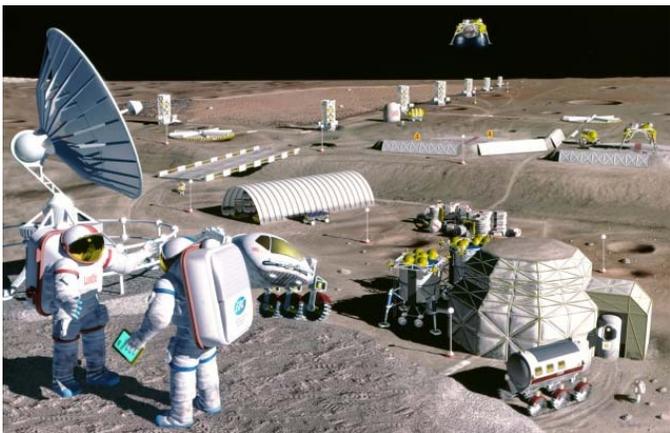


# Specific Technology Focus at JPL

Application	Technical Focus
Automobiles	<ul style="list-style-type: none"><li>• Reducing precious metal loading to 0.2 mg/cm<sup>2</sup></li><li>• Increasing the durability of catalysts to 5000 hours</li></ul>
Lunar Systems	<ul style="list-style-type: none"><li>• Improving Efficiency PEM H<sub>2</sub>/O<sub>2</sub> fuel cell</li><li>• Advanced Oxygen Evolution Catalysts for PEM water electrolysis</li><li>• Electrolysis Stacks capable of balanced pressure operation 2000 psi</li></ul>



# PEM Fuel Cell for Lunar Lander (Altair), Lunar Surface Mobility systems, and Lunar Habitats.



- **Advantages of PEM fuel cells over state-of-art alkaline fuel cells include :**
  - **lower mass and volume**
  - **enhanced safety**
  - **longer life, > 5000 hours**
  - **increased flexibility for load handling**
  - **improved reliability**



# Approach

- Fabrication of Nafion-Based MEAs
  - JPL in-house methods
  - Procured from two other Vendors
  - Evaluate membranes of various thicknesses and EW

## MEA performance characterization

- Operating conditions
- Membrane types and thicknesses
- Backing papers
- Catalyst loading

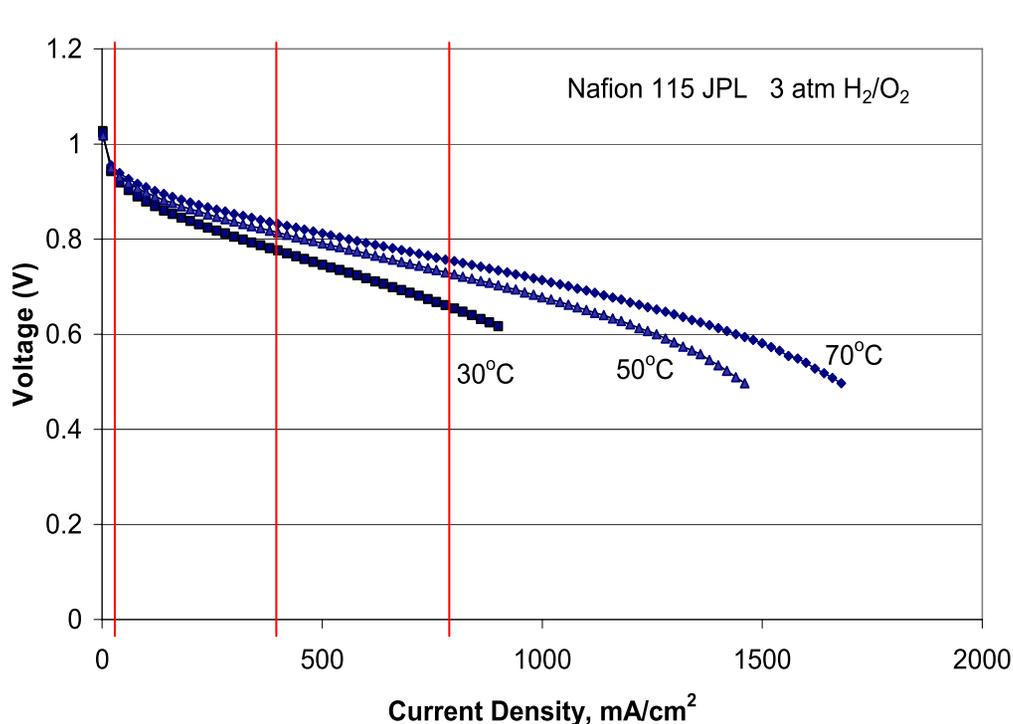
## JPL MEA scale up results

## JPL and vendor MEA comparison



# JPL MEA Performance Characterization

## Effect of Temperature

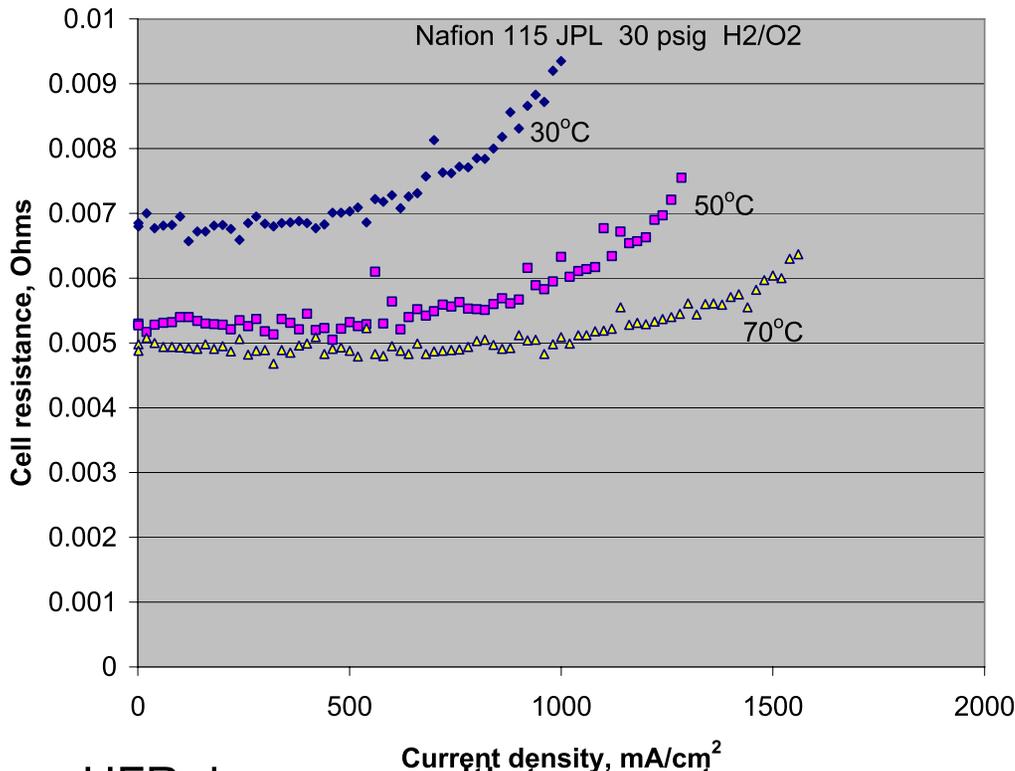


C.D. mA/cm <sup>2</sup>	$\Delta V$ mV
40	25
400	54
800	77

- Performance increase of 20 , 54 and 77 mV at 40, 400 , and 800 mA/cm<sup>2</sup>
- Improvement largely due to ionic resistance of membrane



# Internal Resistance Effect of Temperature



400 mA/cm<sup>2</sup>

°C	HFR mOhm
30	7
50/70	4.5

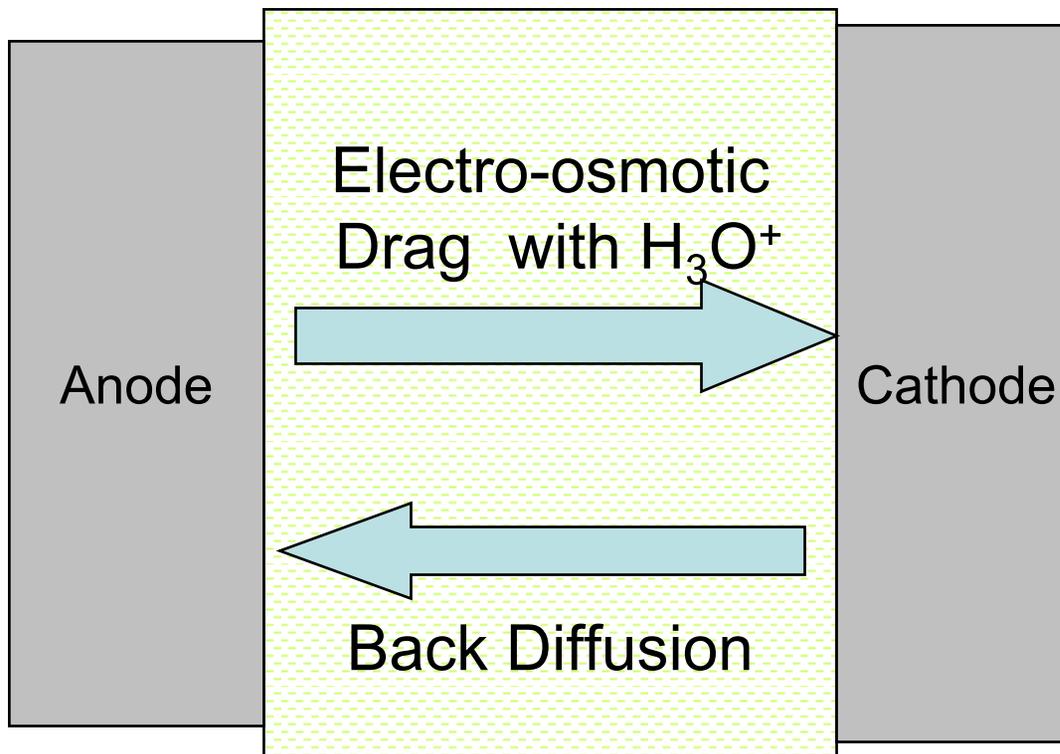
1200 mA/cm<sup>2</sup>

°C	HFR mOhm
30	-
50	5.8
70	5

- HFR decreases with temperature
  - Nafion conductivity increases with temperature
- HFR increases with current density
  - point of increase pushed to higher current density with increase in temperature



# Water Transport Processes in an MEA

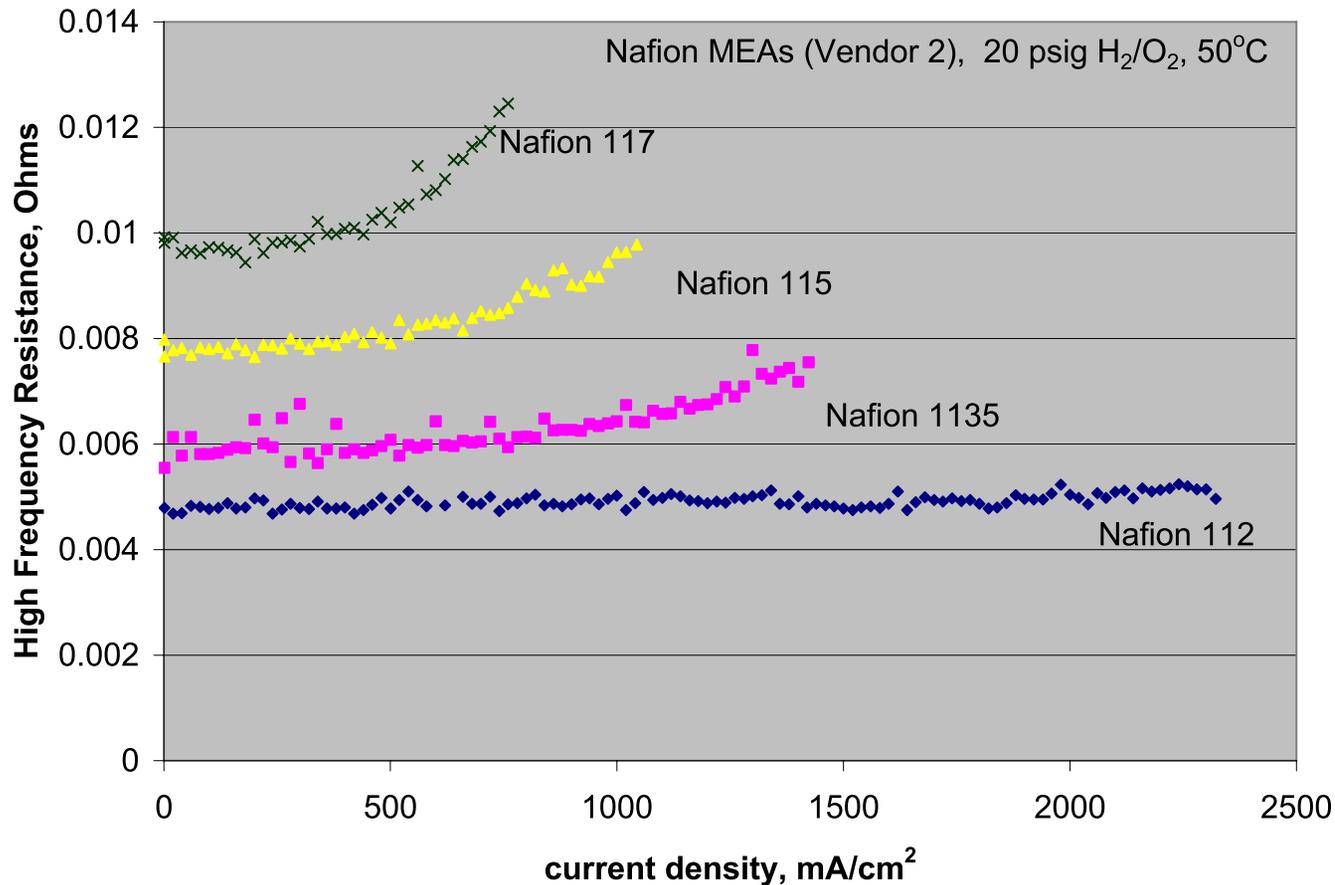


Felix Buchi and Gunther Scherer, *J. Electrochem. Soc.*, **148** A183 (2001).

M. Eikerling, Yu. I. Kharkats, A.A. Kornysheve and Yu. M. Volfkovich, *J. Electrochem.Soc.*, **145** 2684 (1998).



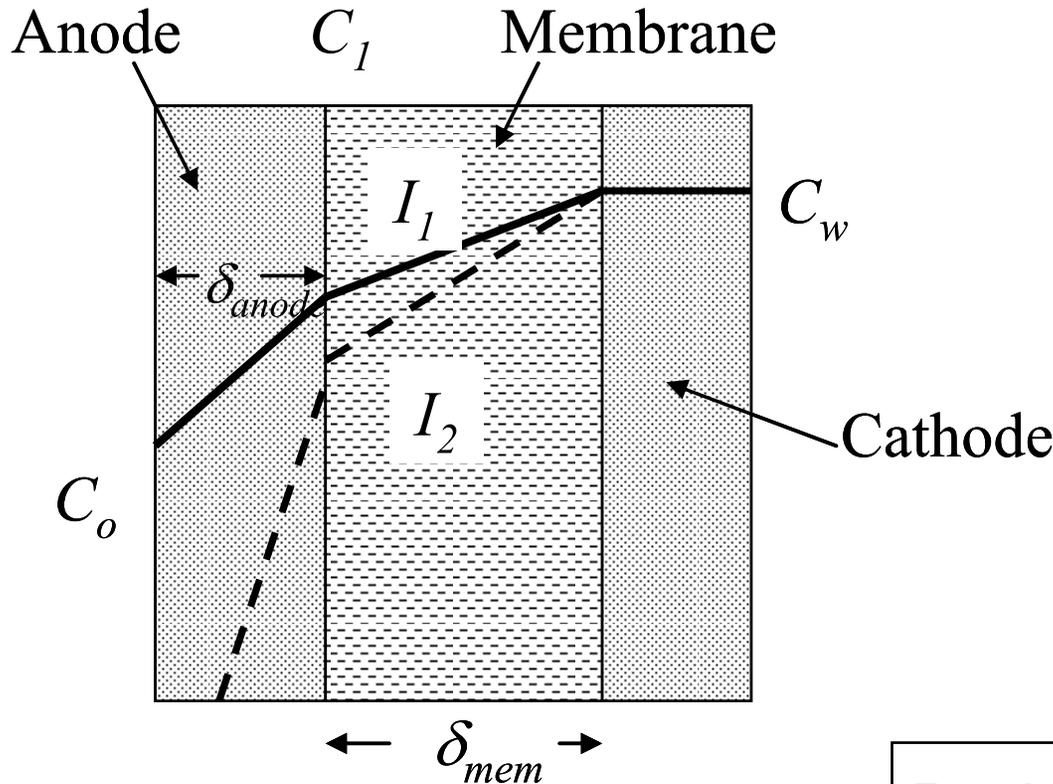
# Membrane Thickness and Resistance



- Thinner membranes exhibit low constant resistance over wide range of current density



# One-Dimensional Analysis of Water Transport



$\gamma$  is the electrosmotic coefficient,  
 $I_{bdl}$  is the current density,  
 $F$  is the Faraday constant,  
 $D_w$  is the diffusion coefficient of water  
 $C_w$  is the concentration at the cathode end of the membrane  
 $\delta_{mem}$  is the thickness of the membrane electrolyte. ,

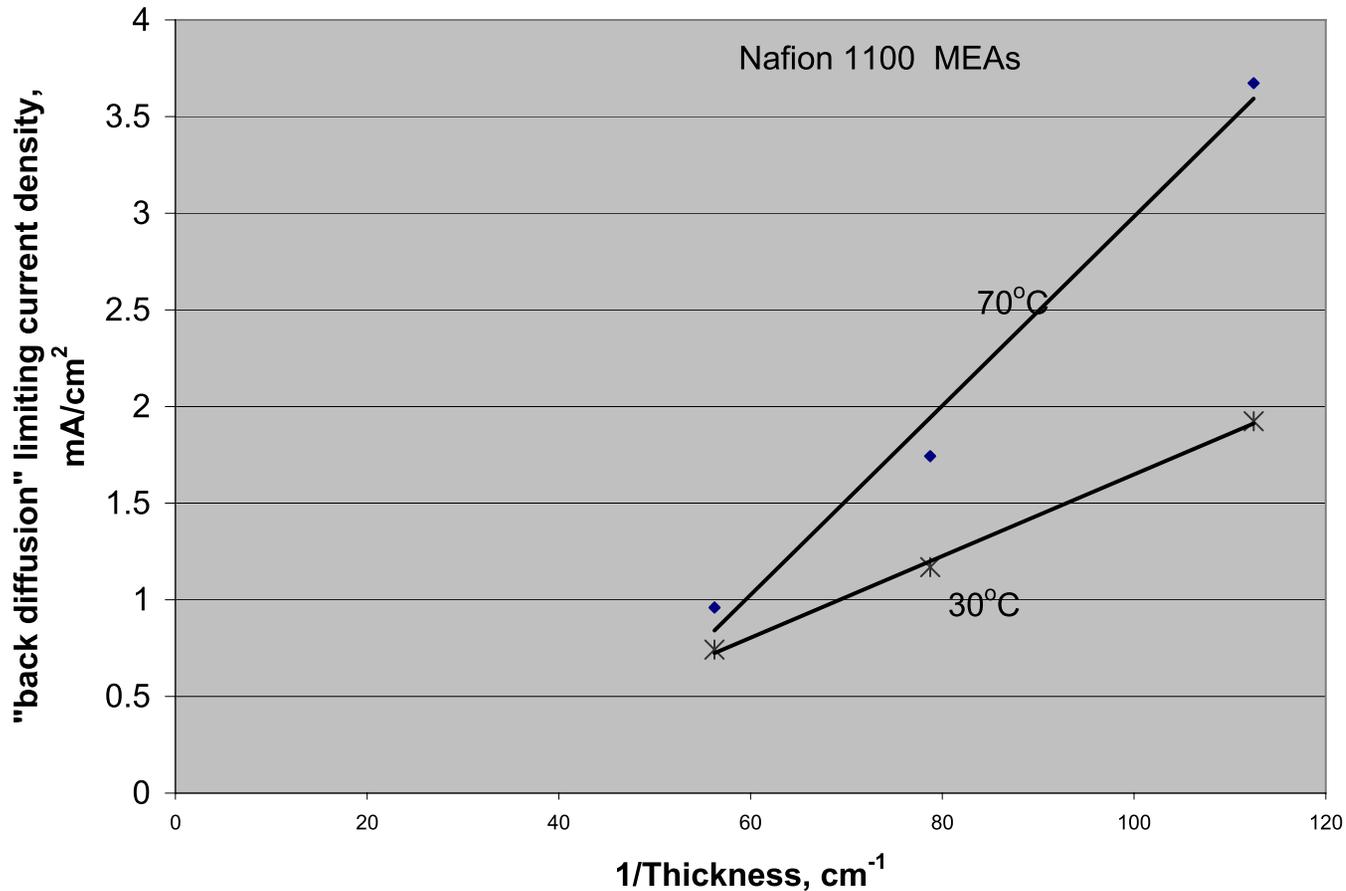
$$\frac{\gamma \cdot I}{F} = \frac{k^o D_w \{C_w(I) - C_1(I)\}}{\delta_{mem}(I)}$$

Back-diffusion Limited current

$$I_{bdl} = \frac{D_w C_w F}{\gamma \cdot \delta_{mem}}$$



# Verification of Analysis



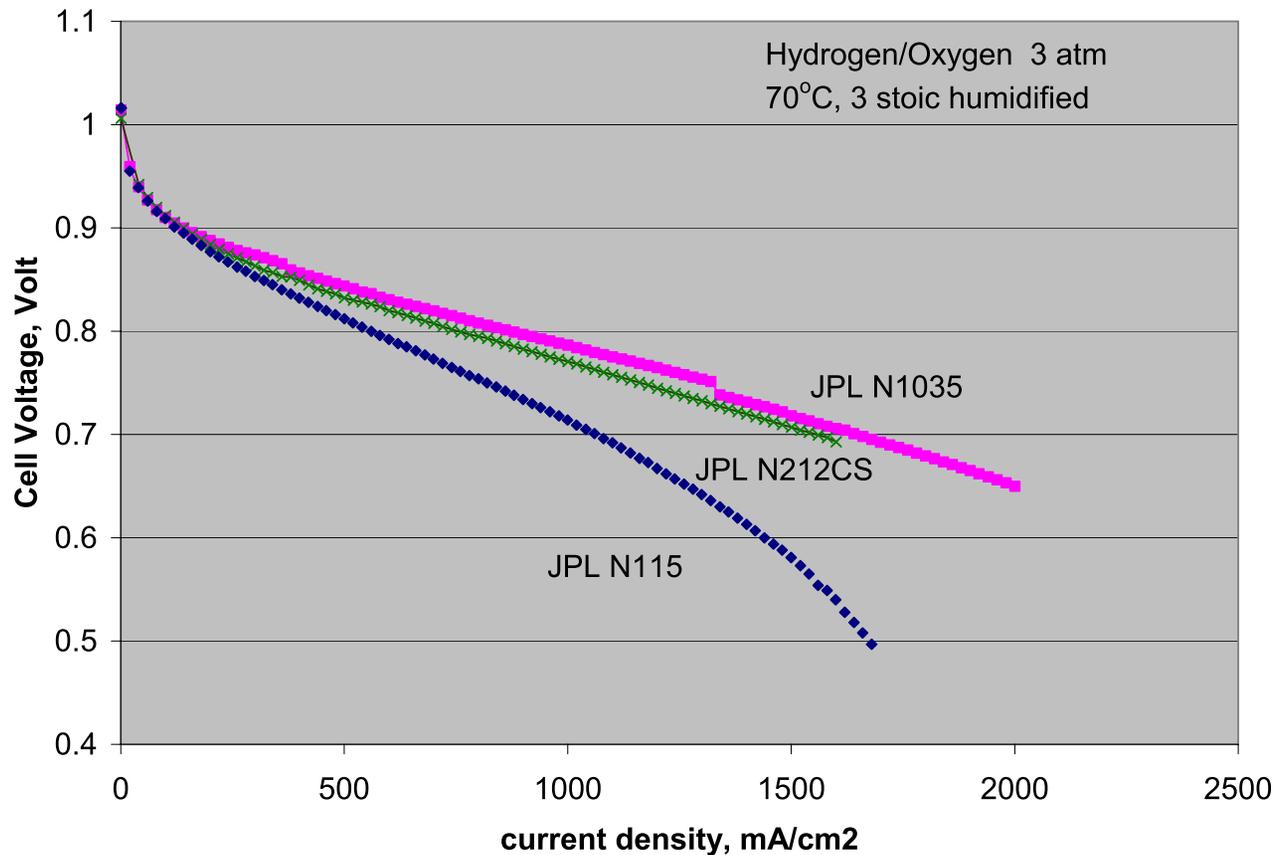
$D_w$  determined to be  $2 \times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$  at 30°C ; consistent with literature data\*

\*Thomas A. Zawodzinski, Jr., Thomas E. Springer, John Davey, Roger Jestel, Cruz Lopez, Judith Valeria, and Shimshon Gottesfeld, *J. Electrochem. Soc.*, Vol. 140,1993 p.1981.



# JPL MEA Performance Characterization

## Nafion 1035, 212CS & 115 JPL MEAs

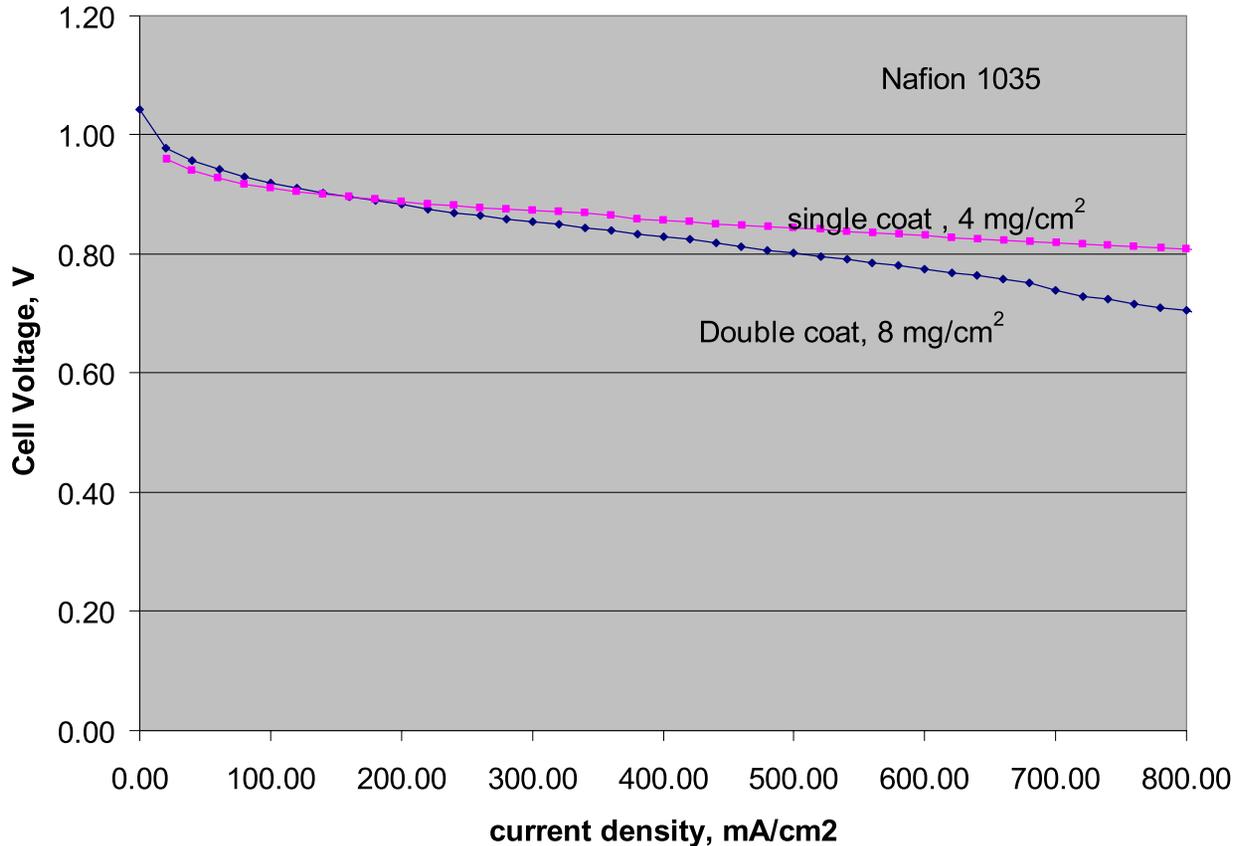


- Performance increases as membrane thickness decreases
- Performance increases as equivalent weight decreases



# Nafion 1035 JPL

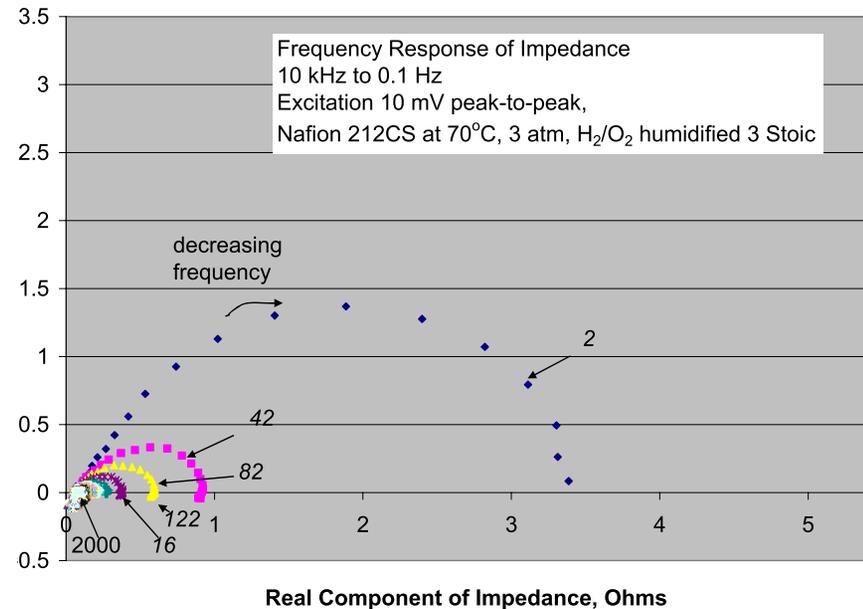
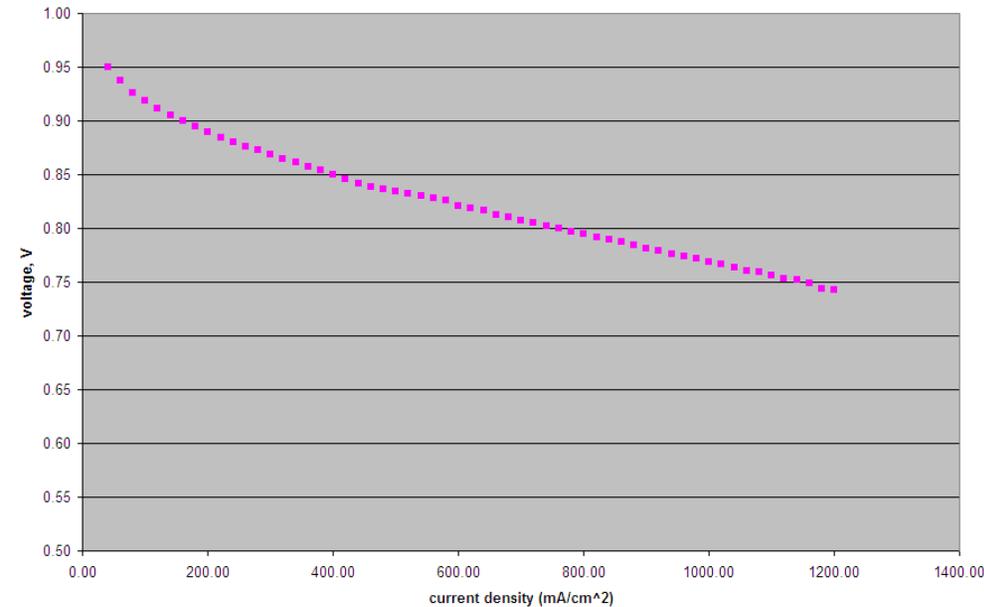
## Comparison of 4 mg/cm<sup>2</sup> and 8 mg/cm<sup>2</sup>



Double coat (8 mg/cm<sup>2</sup>) shows slightly better performance over single coat( 4 mg/cm<sup>2</sup>) at low current densities. The higher loading does seem to reduce performance at higher current densities.



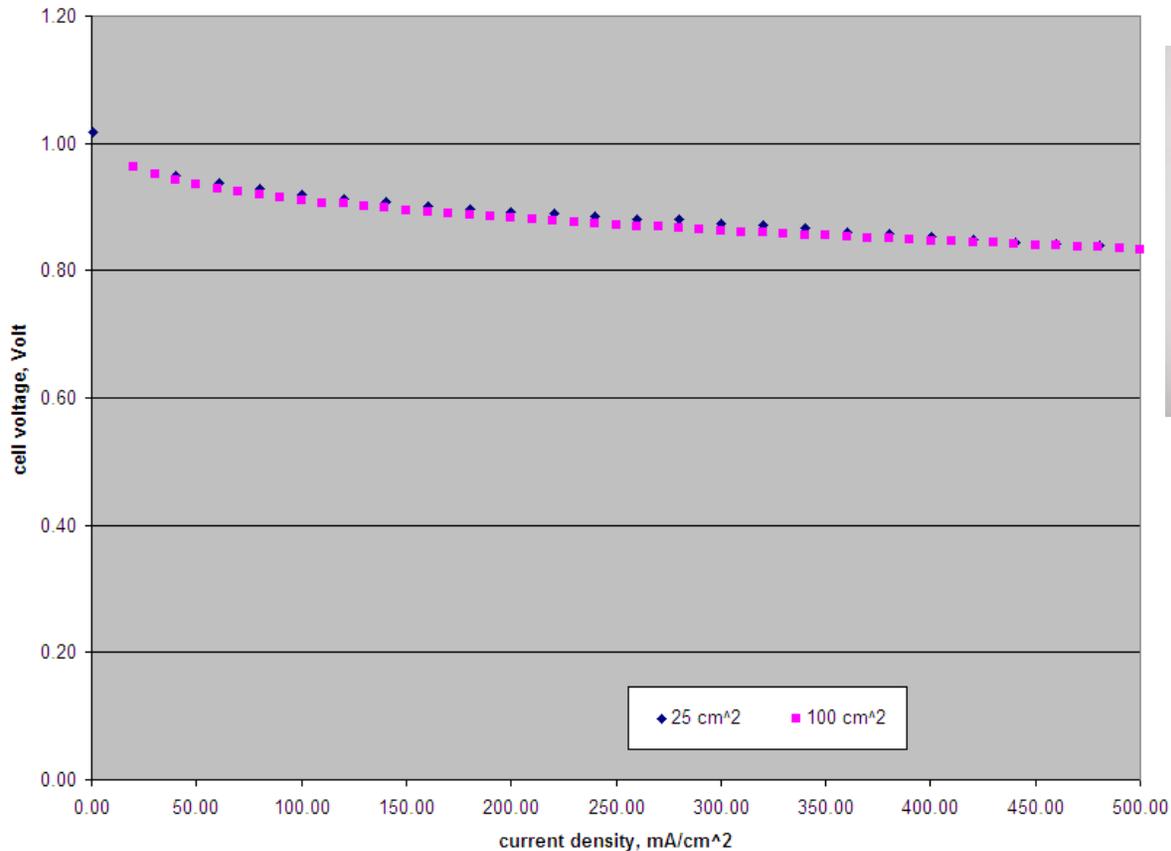
# Polarization Characteristics of - 212CS JPL MEA



- Mass transfer effects not seen at high current densities
- Thinner membrane allows faster back-diffusion to anode



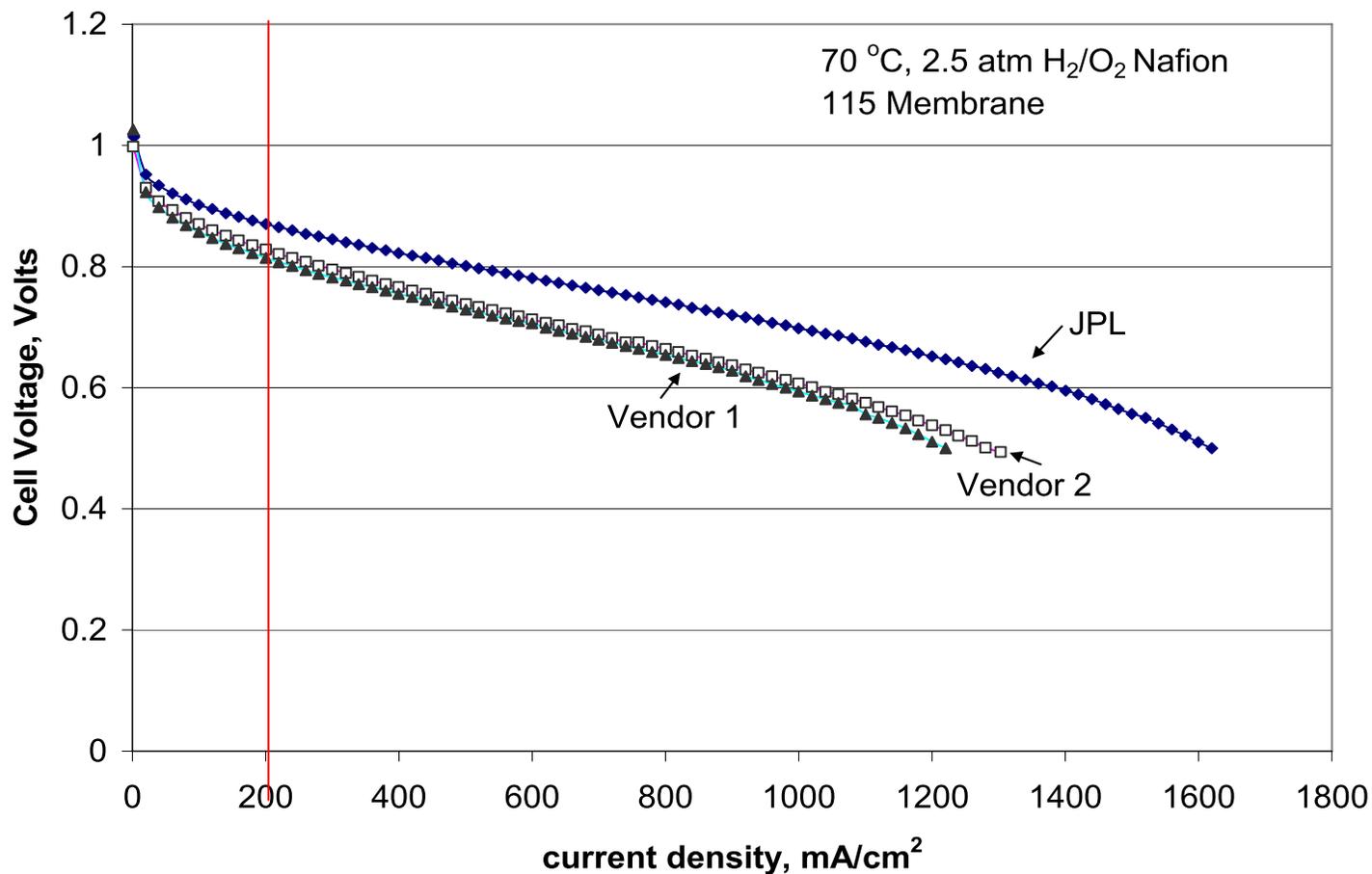
# Nafion 212CS based MEA Scale-up Experiments



- Results on 25 cm<sup>2</sup> and 100 cm<sup>2</sup> active area demonstrates scalability of MEA manufacturing process loss in performance



# JPL and Vendor MEA Performance Comparison



- JPL MEA ~ .88 V at 200 mA/cm<sup>2</sup>
- Vendor MEAs ~ .82-.84 at 200 mA/cm<sup>2</sup>



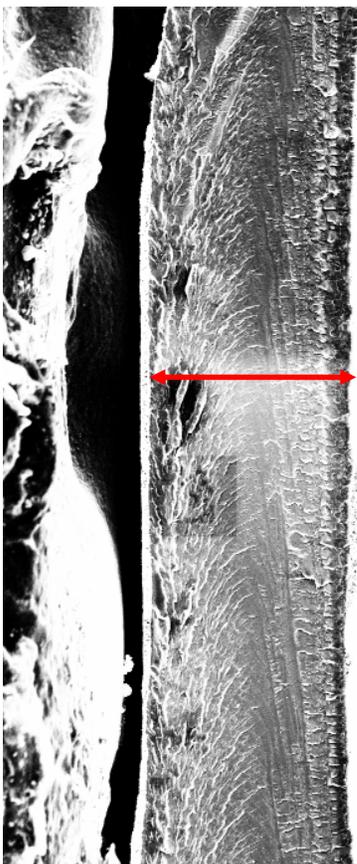
# JPL and Vendor MEA Performance Comparison

MEA Type	Processing	Gas Diffusion Layer
N115 JPL	Paint / Hot Press	Catalyst coated Toray 060 / Hot Pressed and bonded
N115 Vendor 1	Screen Print / Cold Press ?	Uncatalyzed carbon cloth / Felt; not bonded
N115 Vendor 2	Transfer / Hot Press?	Carbon coated Toray 060; Non-bonded



# SEM cross-sections of JPL & Vendor MEAs

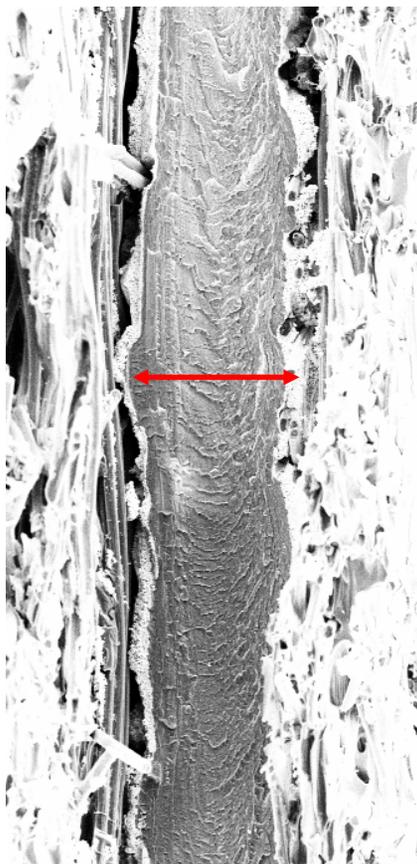
Vendor 1



Signal A = SE2  
Mag = 500 X

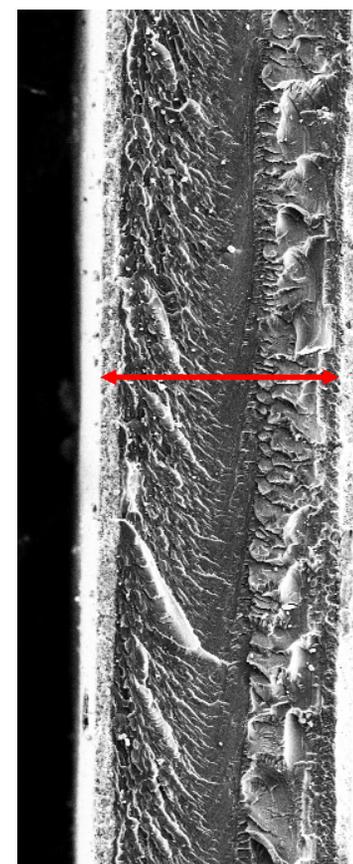
D  
Ti

JPL



Signal A = SE2  
Mag = 500 X

Vendor 2



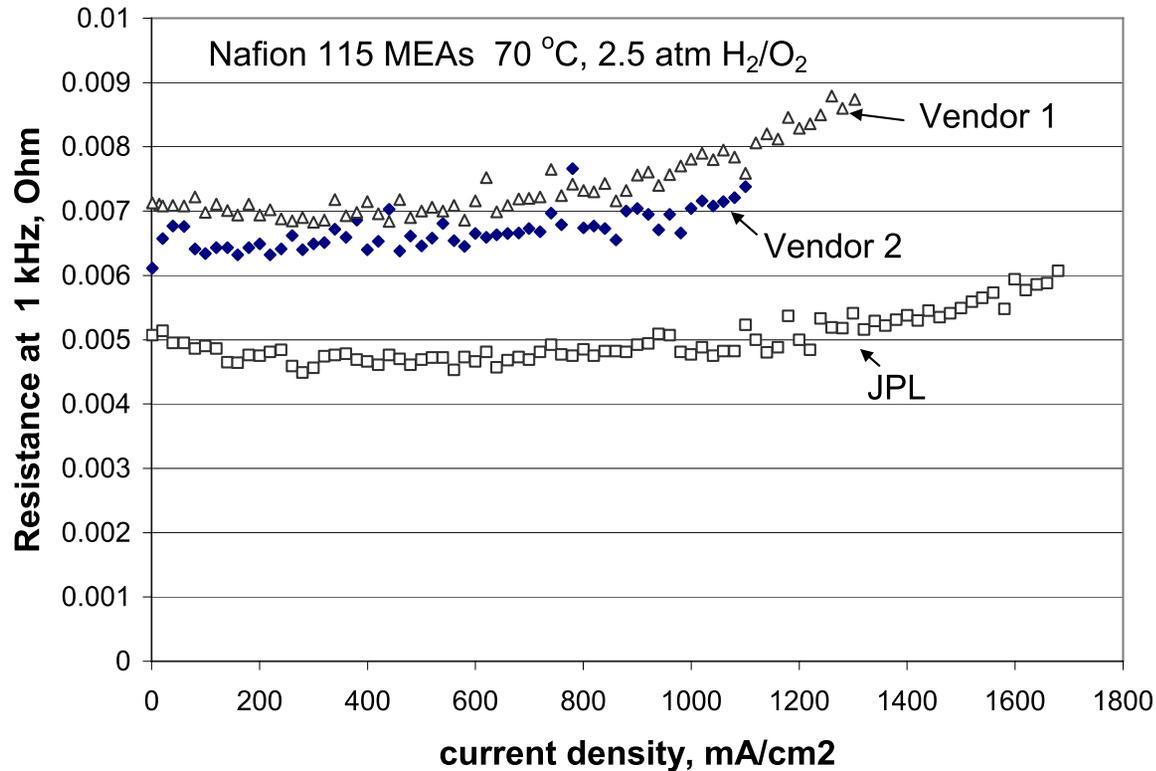
Signal A = SE2  
Mag = 500 X

Date :1 C  
Time :12

JPL process results in about 10-15% reduction in membrane thickness



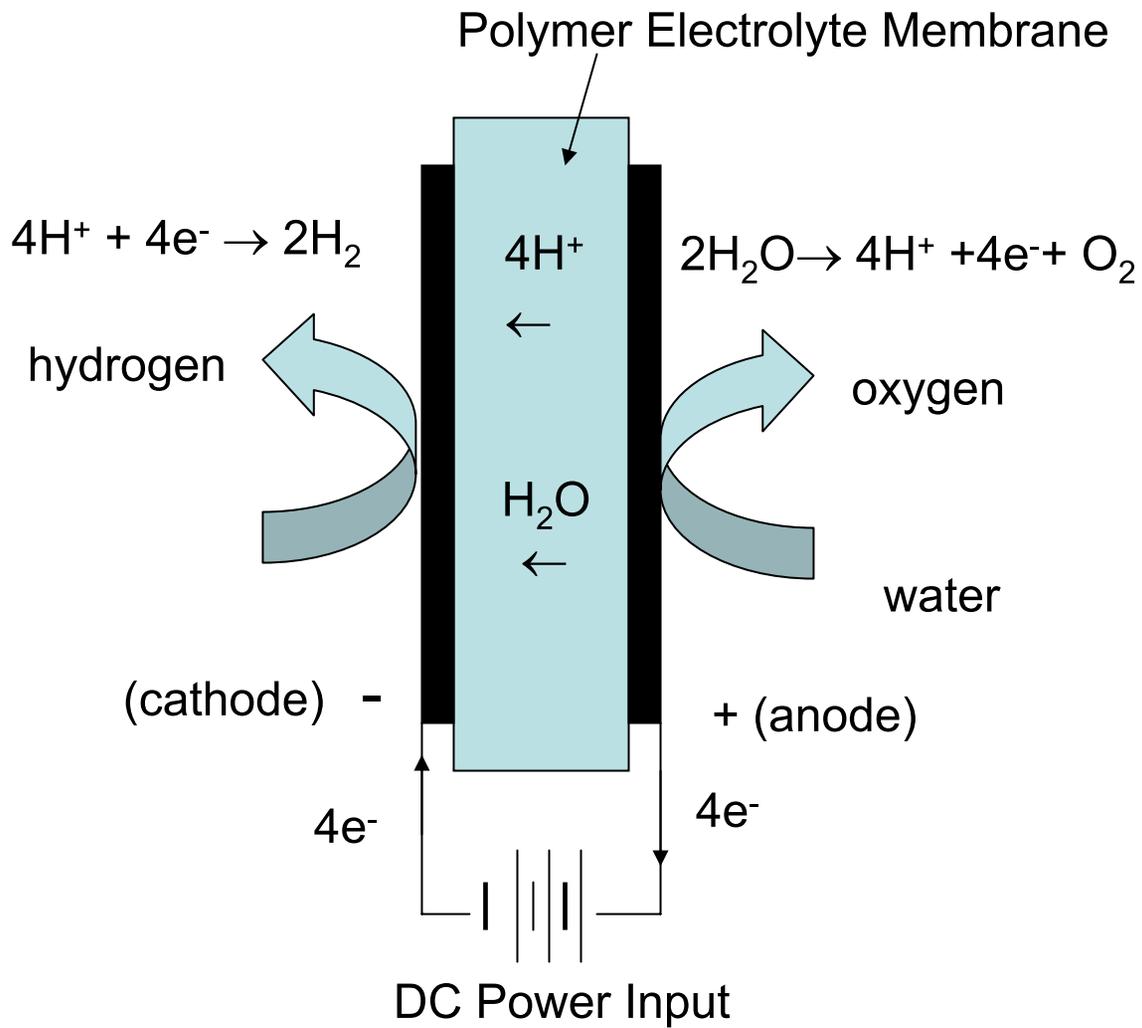
# JPL and Vendor MEA Performance Comparison



- JPL MEA ~ 5 milliohms,
- Vendor 1 ~6.5 milliohms,
- Vendor 2 ~ 7 milliohms

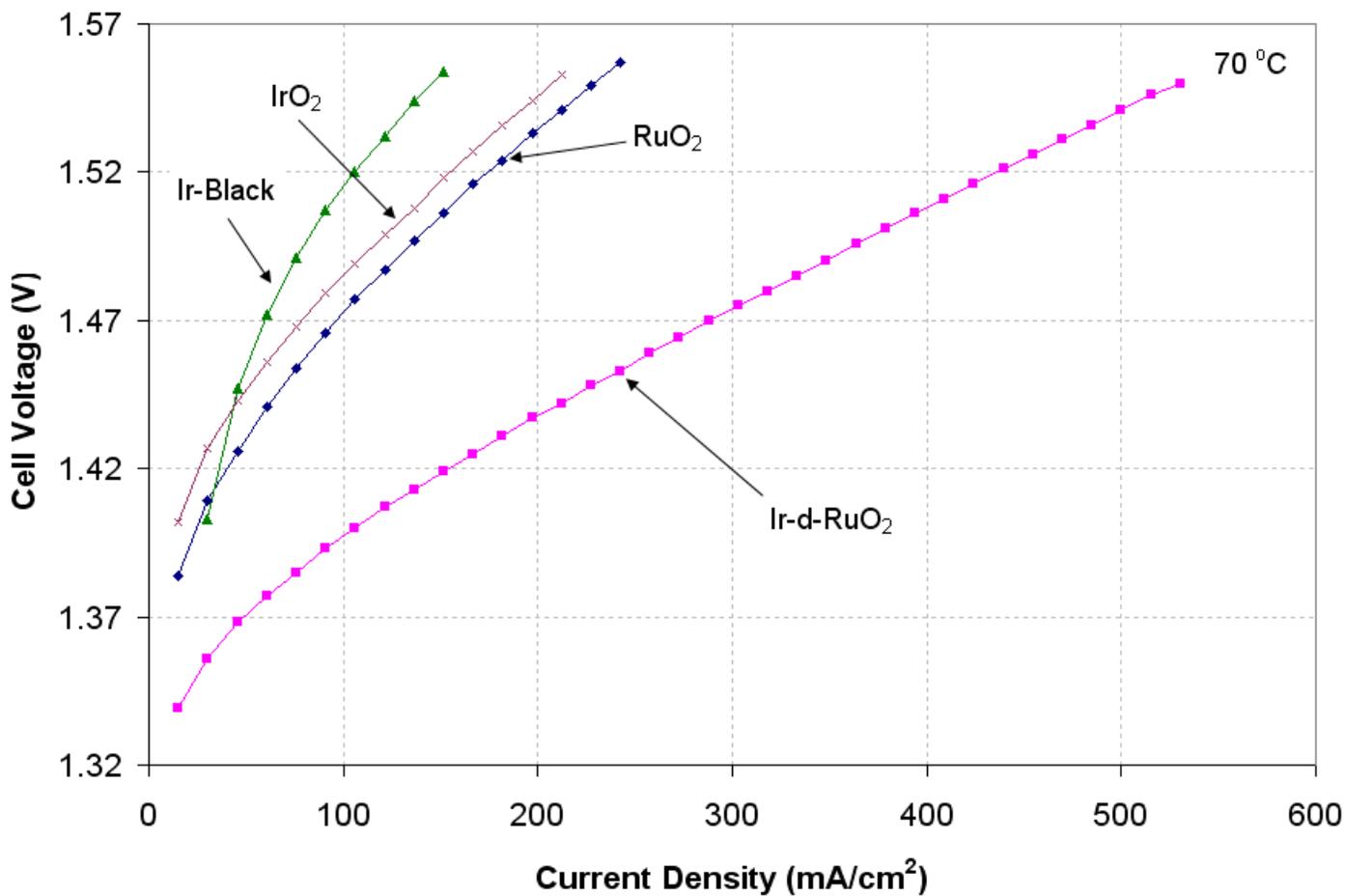


# Schematic of Water Electrolysis





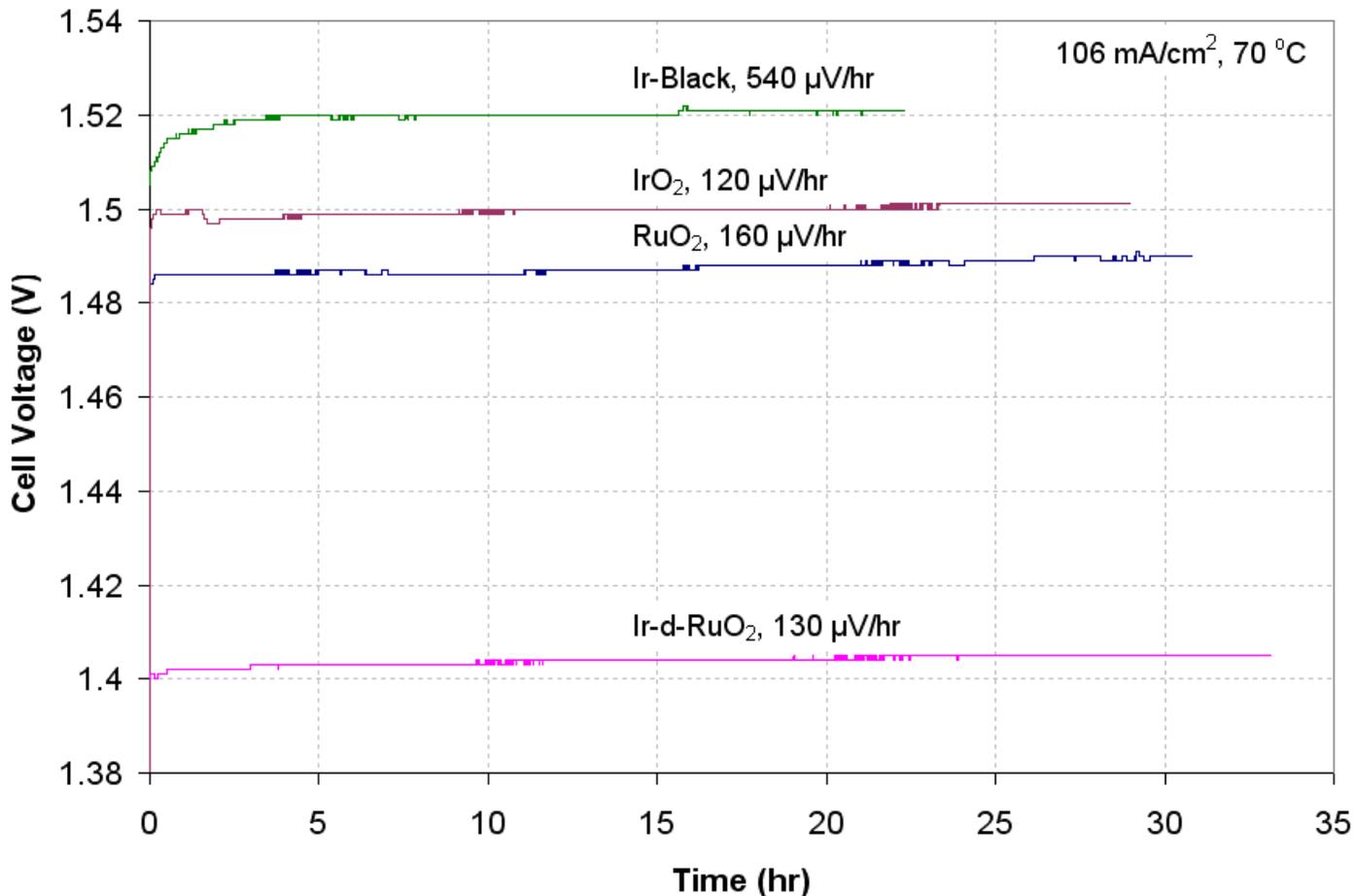
# Iridium-doped Ruthenium Oxide Catalyst Performance Versus Commercially Available Materials



- The iridium-doped ruthenium oxide catalyst performed better than the individual constituents



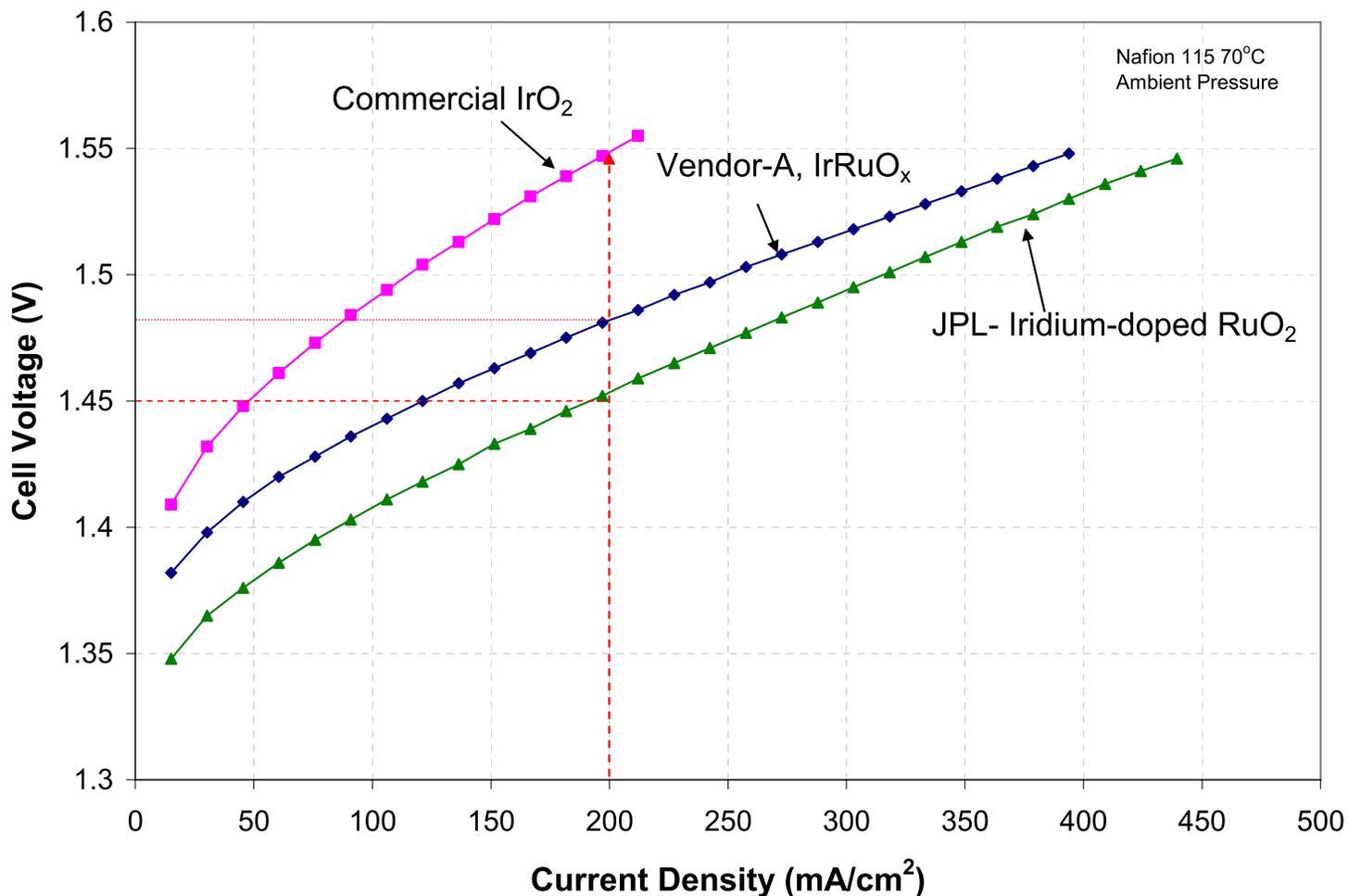
# Short-Term Durability Tests



- Stability testing exhibit the same performance trends as polarization experiments
- Ir-doped ruthenium oxide catalyst as stable as iridium oxide



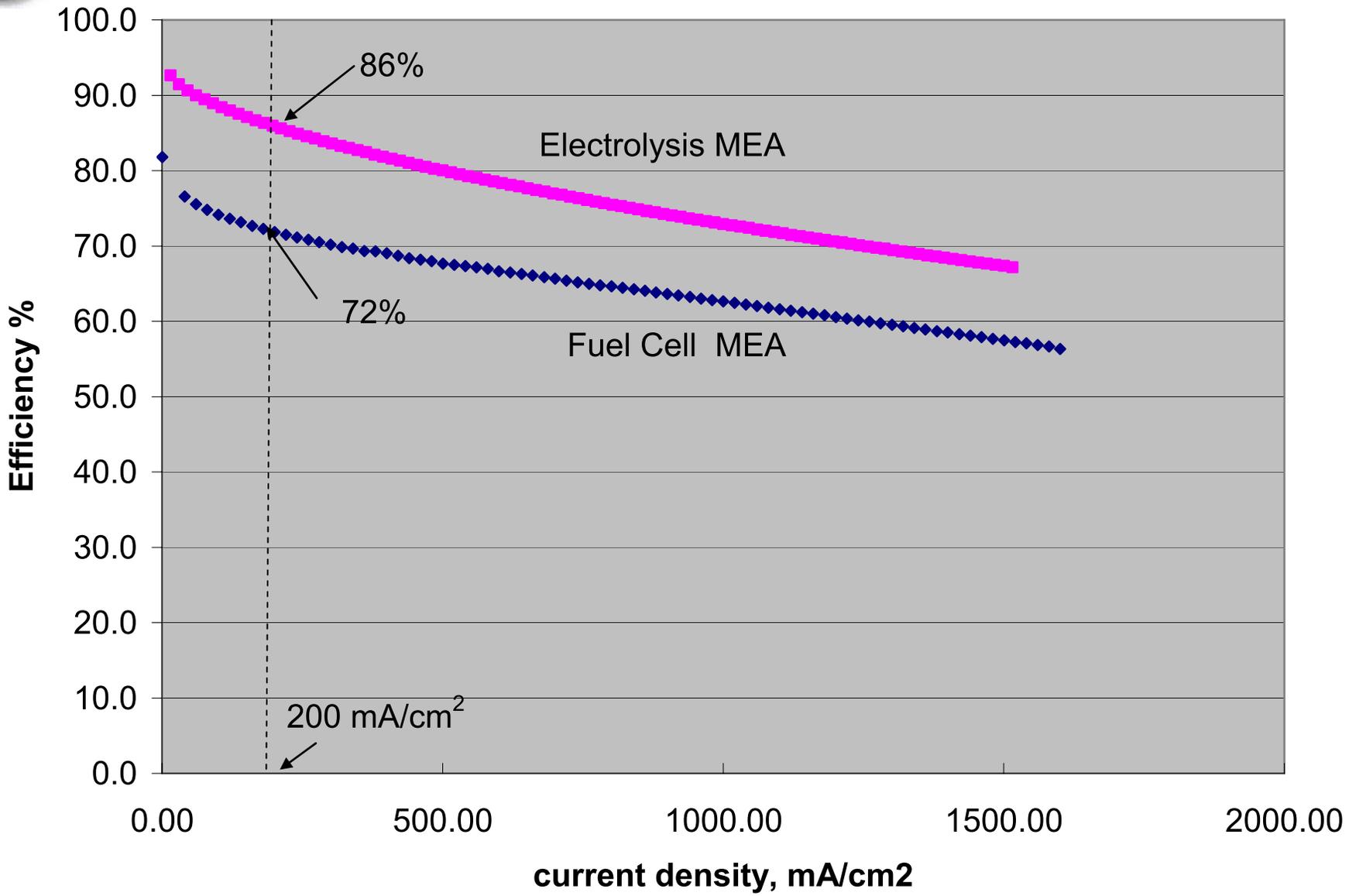
# Advances in Electrolysis MEAs at JPL



- JPL's Iridium-doped Ruthenium oxide catalyst exhibits higher performance compared to commercial and vendor-supplied materials.
- With the performance of 1.45V at 200 mA/cm<sup>2</sup> these JPL MEAs have the potential of meeting the full success criteria



# Status of the Performance of Advanced MEAs from JPL



Demonstrated Performance exceeds minimum success criteria for MEAs



# Acknowledgement

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